

electron energy and light pdf

Electron Energy and Light 5 20. Label the remaining drawings in Model 3 with the electron transitions that are occurring ($n=?$ to $n=?$), the wavelengths and corresponding colors as given in example A in Model 3.

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(b) This electron moves from a lower / higher energy state to a lower / higher energy state. (c) Light is absorbed / released in the 3 to 2 electron transition. 23.

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This electron moves from a (lower/higher) energy state to a (lower/higher) energy state. b. Inc. Identify the drawing in Model 3 that depicts a hydrogen atom with an electron moving from energy level 5 to energy level 2.

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