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### **Holt Chemistry Section Intermolecular Forces**

Dipole-Dipole Interactions or Dispersion Forces If two molecules are of comparable size and shape, dipole-dipole interactions will likely be the dominating force. If one molecule is much larger than another, dispersion forces will likely determine its physical properties.

### **Chapter 11 Intermolecular Forces - Department of Chemistry**

Holt Chemistry Ch. 11 States of Matter and Intermolecular Forces. chemistry. STUDY. PLAY. surface tension. a force that acts on the surface of a liquid and that tends to minimize the area of the surface. evaporation. the process by which water changes from liquid form to an atmospheric gas.

### **Holt Chemistry Ch. 11 States of Matter and Intermolecular**

What force is there between oppositely charged objects? SAFETY PRECAUTIONS CONTENTS SECTION 1 States and State Changes SECTION 2 Intermolecular Forces SECTION 3 Energy of State Changes SECTION 4 Phase Equilibrium 1 2 3 11 START-UPACTIVITY 377 Heating Curve for Water PROCEDURE 1. Place several ice cubes in a 250 mL beaker. Fill the beaker halfway with water.

## **CHAPTER**

Intermolecular forces- forces of attraction and repulsion between molecules that hold molecules, ions, and atoms together. Intramolecular - forces of chemical bonds within a molecule Boiling Point and Electronegativity Boiling Point - the temperature at which the liquid form of a compound vaporizes into a gas.

### **INTERMOLECULAR FORCES - UCLA Chemistry and Biochemistry**

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### **Holt Chemistry Chapter 11: States of Matter and**

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