

### **naming ionic compounds practice pdf**

Review Naming Chemical Compounds The following are a good mix of naming and formula writing problems to help you get some practice. Name the following chemical compounds: 1) NaBr \_\_\_\_\_ 2)  $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$  \_\_\_\_\_ 3)  $\text{P}_2\text{O}_5$  \_\_\_\_\_

### **Naming Ionic Compounds Practice Worksheet**

CHEMISTRY Naming Compounds Handout page 1 of 12 IONIC COMPOUNDS versus MOLECULAR COMPOUNDS ionic compound: consist of cations (positive ions) and anions (negative ions) held together by electrostatic attraction

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For the following compounds, give the formulas 22) sodium phosphide \_\_\_\_\_ 23) magnesium nitrate \_\_\_\_\_ 24) lead (II) sulfite \_\_\_\_\_

### **Naming Ionic Compounds Worksheet I - IMSA**

naming ionic compounds practice pdf In chemistry, an ionic compound is a chemical compound composed of ions held together by electrostatic forces termed ionic bonding. The compound is neutral overall, but consists of positively charged ions called

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Many compounds exhibit predominantly either ionic or covalent bonding. ... of compounds include water, which is a combination of the elements oxygen and ... NAMING AND WRITING CHEMICAL COMPOUNDS write the name of the negative ion. - cross over the charges to write the formula eg.

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Naming Ionic Compounds Practice Worksheet - Solutions Name the following ionic compounds: 1)  $\text{NH}_4\text{Cl}$  ammonium chloride 2)  $\text{Fe}(\text{NO}_3)_3$  iron (III) nitrate 3)  $\text{TiBr}_3$  titanium (III) bromide 4)  $\text{Cu}_3\text{P}$  copper (I) phosphide 5)  $\text{SnSe}_2$  tin (IV) selenide 6) GaAs gallium arsenide

### **Naming Ionic Compounds Practice Worksheet - Solutions**

Practice naming ionic compounds when given the formula If you're seeing this message, it means we're having trouble loading external resources on our website. If you're behind a web filter, please make sure that the domains \*.kastatic.org and \*.kasandbox.org are unblocked.

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Chapter 7 Ionic Compound Naming (Practice Quiz) (with oxidation numbers and correct subscript latex codes)

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