

### Paula Bruice Solution PDF

Looking at the reaction profile of EAS, you will see the energy required for step 1 to proceed is very large. Like SN1 and E1 reactions, the rate-determining step (RDS) is the formation of the carbocation. Looking back at the three carbocation fates (capture a nucleophile, lose a proton and form a pi bond, or

### Electrophilic Aromatic Substitution (EAS) is a

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Hydrogen iodide (HI) is a diatomic molecule and hydrogen halide. Aqueous solutions of HI are known as hydroiodic acid or hydriodic acid, a strong acid. Hydrogen iodide and hydroiodic acid are, however, different in that the former is a gas under standard conditions, whereas the other is an aqueous solution of said gas.

### Hydrogen iodide - Wikipedia

A carbocation (/ ˈkɑːbəkəˈɪʃən / or / ˈkɑːbəkəˈɪʃən /) is an ion with a positively charged carbon atom. Among the simplest examples are the methenium CH<sub>3</sub><sup>+</sup>, methanium CH<sub>5</sub><sup>+</sup> and vinyl C<sub>2</sub>H<sub>3</sub><sup>+</sup> cations. Occasionally, carbocations that bear more than one positively charged carbon atom are also encountered (e.g., ethylene dication C<sub>2</sub>H<sub>2</sub><sup>2+</sup>).

### Carbocation - Wikipedia

En chimie, l'équation de Henderson-Hasselbalch est une équation donnant le pH d'un système tamponné. Elle est notamment utilisée en physiologie et en médecine pour déterminer le pH sanguin à partir des concentrations en ion bicarbonate et en acide carbonique.

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