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## Lab 4 Intermolecular Forces Pcc Answer Key

**lab 4 intermolecular forces - west windsor-plainsboro ...** - lab 4 name \_\_\_\_\_ intermolecular forces pre-lab assignment this written pre-lab is worth 3 points of your lab report grade and must be turned in to your lab instructor before class begins. 1. read the entire lab handout. 2. make a table listing the names and chemical formulas of all chemicals used in this lab. for organic **laboratory: intermolecular forces (imf) - parkland college** - laboratory: intermolecular forces (imf) report requirement: answer all of the questions/do all the computations requested in italics. questions not in italics do not need to be answered. you do not have to write a formal lab report. you should write your answers into a word processing program and save the file. go into the lab-imf report and ... **intermolecular and ionic forces - welcome to web.gccaz** - gcc chm 151II: intermolecular and ionic forces ©gcc, 2013 page 4 of 12 intermolecular and ionic forces comparing imf using percent evaporation: this lab will explore changing masses during evaporation of organic liquids. the decrease in mass is related to the strength of the intermolecular forces of attraction. consider the strength of the forces **experiment: evaporation and intermolecular attractions** - post lab questions (to be answered in your lab report) 1. what are intermolecular forces and how do they affect physical states? 2. which of the alcohols (first 4 substances tested) studied had the strongest intermolecular forces of attraction? the weakest intermolecular forces? explain using the results of this experiment. 3. **lab # 10: intermolecular forces of attraction lab** - lab # 10: intermolecular forces of attraction lab accelerated chemistry i there are different kinds of intermolecular forces of attractions: dispersion forces, dipole-dipole interactions and hydrogen bonds (which are a special type of dipole-dipole interaction). **intermolecular forces lab - annville-cleona school district** - intermolecular forces lab research questions how does a compound's molecule mass affects its boiling point? how do a compound's intermolecular forces affect its boiling point? what are the relative strengths of the four intermolecular forces (imfs)? compounds butane (c 4 h 10) chlorine trifluoride (clf 3) chloromethane (ch 3 cl) ethane (c **lab 4 intermolecular forces pcc answer key** - intermolecular forces (imf) ... compounds can burn we focus in this lab on the hydrocarbons. ... 4 (water soluble) + saturated hydrocarbons are unreactive advising handbook - north carolina state university **intermolecular forces of various substances lab report** ... - point of different liquids. intermolecular forces affect the boiling point of substances because, like the melting point, one is loosening those intermolecular bonds. in the case of the new, hypothetical lab, the boiling point is just taking it another step further than the melting point from this lab.

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